CHEM 312



THIRD YEAR EXAMINATION FOR THE AWARD OF THE DEGREE OF BACHELOR OF EDUCATION SCIENCE FIRST SEMESTER 2021/2022 (JULY, 2022)

CHEM 312: ALCOHOLS, ETHERS AND CARBONYLS

STREAM: Y3 S1

TIME: 2 HOURS

DAY: FRIDAY, 12.00 PM - 2.00 PM

DATE: 00/07/2022

INSTRUCTIONS:

- 1. Do not write anything on this question paper.
- 2. Answer ALL Questions in section A and any TWO Questions in Section B.

SECTION A

- 1. (a) Explain is meant by hydroboration-oxidation(2 marks)
 - (b) Give the structures of the of the products in the following reactions

(i)
$$H_{3C} \xrightarrow{CH_{3}}_{C} = CHCH_{3} \xrightarrow{(i) \text{ THF, BH3}}_{(ii) \text{ H2O}_{2}, \text{ OH}^{-}}$$

(ii) $H_{3C} \xrightarrow{O}_{S} = CI + H = OCH_{2}CH_{3} \xrightarrow{Base, -HCI}$

2. Draw the structure and give the IUPAC name of the products in the given reactions

(4 marks)

(2 marks)

- (i) H_2C H^+, H_2O (ii) H^+, H_2O
- 3. Show the mechanism for the reaction below that shows the synthesis of alcohols (6 marks)



4. (a) Starting with 2-methyl-2-butene, show all the steps involved in the synthesis of 2methyl-2-butanol (4 marks)

(b) Draw the structures of the major products labeled A and B in the following reactions involving alcohols (2 marks)



5. Williamson synthesis synthesis of ethers involves a reaction between sodium alkoxide with an alkyl halide, alkyl sulphonate or an alkyl sulphate. Show the reaction mechanism for the synthesis of ethyl propyl ether from propyl alcohol (6 marks)

$$CH_{3}CH_{2}CH_{2}OH \xrightarrow{(i) \text{ NaH}} CH_{3}CH_{2}OCH_{2}CH_{3}$$

$$(ii) CH_{3}CH_{2}I \xrightarrow{(ii) CH_{3}CH_{2}I} 6. The two$$

synthetic routes for 2-ethoxy-1-phenylpropane is outlined below. Draw the structure of the products shown by letters A, B, C and D (4 marks)



7. Epoxidation of cyclopentene produces 1, 2-epoxycyclopentane. Give the reaction mechanism for this reaction (6 marks)



8. Using suitable examples, distinguish between protic and aprotic solvents (4 marks)

SECTION B

9. (a) Give the IUPAC names for the following organic compounds (5 marks)



(b) Give the products in the following reactions. (5 marks)



- (iii) OH HCI (ii)
- (c) Show the reaction mechanism in the following reaction (7 marks)
 CH₃CH₂OH + CH₃CH₂OH → H₂SO₄ → CH₃CH₂OCH₂CH₃
 10. (a) The oxo-process provides a synthetic route for primary alcohols. Describe the synthesis of propanol starting with ethene using the oxo-process (9 marks)
 - (b) Explain the advantages of the oxo-process over other methods of synthesis (6 marks)
- 11. (a) Describe the manufacture of the following solvents in the industrially
 - i) Diethyl ether (DEE) (5 marks)
 - (ii) Ethylene glycol (5 marks)
 - (b) Explain any five limitations associated with polar aprotic solvents during in chemical reactions (5 marks)