KISII UNIVERSITY

FIRST-YEAR EXAMINATION FOR THE AWARD OF THE DEGREE OF

MASTER OF SCIENCE (PHYSICS)

(January 2022-May 2022)

PHYS 824: ATOMIC AND MOLECULAR PHYSICS

INSTRUCTIONS

Answer question one and any other two questions in the answer booklet

provided.

Duration: 2 hours

QUESTION ONE

(a)	Explain the Pauli Exclusion Principle.	2marks]
(b)	Describe Zeeman Effect and Paschen-Back Effect of an electron sy	stem. [4marks]
(c)	Explain significance of Fortrat Diagrams in structural analysis of a	naterials. 2marks]
(d)	Discuss differences between atomic electronic spectra and molecu electronic spectra.	lar [2marks]
(e)	Calculate the value of the rotational constant B for a molecule of c monoxide. The bond length (R) of CO is 0.113 nm.	arbon [4marks]
(f)	Explain the term Compton scattering.	[2marks]
(g)	State four quantum numbers used to describe quantum states of electron in an atom.	an [4marks]
(h)	Distinguish between symmetric top and asymmetric top molecules	[2marks]
(i)	Define the term electron configuration.	[2marks]
(j)	Describe two applications of microwave spectroscopy.	[4marks]

(k) Define the term predissociation.

QUESTION TWO

(a) Distinguish between Microwave Spectroscopy and Infra-red Spectroscopy.

(b) Differentiate between rigid and non-rigid rotators.	[2marks]
(c) Explain two applications of Raman scattering.	[4marks]
(d) Describe an equation that represents energy of ejected pho	otoelectron during

- photoelectric effect. [3marks]
- (e) With suitable examples, discuss any two groups of rigid rotors. [4marks]

OUESTION THREE

(a)	Define the term nuclear magnetic resonance.	[2marks]
(b)	Explain two applications of nuclear magnetic resonance.	[4marks]
(c)	Highlight two processes by which photons interact with the matter appreciable energy transfer.	without [4marks]

(d) Write an expression for the moment of inertia of a polyatomic molecule.

[2marks]

(e) Explain what happens when an atom is placed in an external magnetic field (assume the magnetic field is constant in time and direction). [3marks]

QUESTION FOUR

(a) State three techniques used for molecular structure analysis. [3marks]

- (b) Discuss two applications of Infra-red Spectroscopy. [4marks]
- (c) From microwave spectroscopy, bond lengths can be determined with a correspondingly high precision. From the rotational microwave spectrum of ¹H³⁵Cl, we find that B = 10.59342 cm⁻¹. Given that the masses of ¹H and ³⁵Cl are 1.0078250 and 34.9688527 amu, respectively, determine the bond length of the ¹H³⁵Cl molecule. [4marks]
- (d) Differentiate between Main group elements and Transition elements of the periodic table. [2marks]

[2marks]

[marke]

(e) What do you understand by the term spin-orbit interactions? [2marks]